Why Replace BPA?

By Steve Hentges, Ph.D.

It’s no secret that bisphenol A (BPA) has been controversial for quite some time and has received intense scrutiny from scientists, government agencies, the media, and environmental groups. Such scrutiny has resulted in demand from consumers for alternatives to products that contain BPA, and hasty efforts from some manufacturers to provide “BPA-Free” products. While the controversy surrounding BPA may be interesting and important, the controversy by itself does not provide an answer to the key question addressed in this article – Why replace BPA?

There are many sound business reasons to consider alternatives to any material that is currently being used. For example, alternative materials might provide better performance or be easier to process, have better commercial availability or lower cost, or be safer. These factors and others can and do provide the necessary incentive for companies to research and develop alternatives as a routine course of business. But in each case, hard facts should drive the decision, not emotions driven by public controversy. Basing decisions on emotions and controversy alone risks “regrettable substitutions” that compromise performance and/or safety.

This article will focus on two of these factors that are of particular importance for BPA – performance and safety. While compromises might be acceptable for some factors such as cost and commercial availability, performance and safety should be held to the highest standards. In both cases, these factors represent large hurdles to overcome before alternatives to BPA should be seriously considered.

BPA Has Been Used for Decades Because It Performs Well

The two primary uses for BPA, accounting for about 95% of all BPA produced, are to make polycarbonate plastic and epoxy resins. Both are high-performing materials that have been increasingly used for about five decades in a wide range of consumer and industrial applications. Although a variety of factors can influence selection of materials, the exceptional performance of these materials is certainly the primary factor that accounts for their widespread use.

Polycarbonate plastic is a highly shatter-resistant, lightweight and optically clear plastic that also has high heat resistance and excellent electrical resistance. It is this combination of strong attributes that makes polycarbonate unique among the wide range of commercially available thermoplastics and, accordingly, so widely used. For example, common products that take advantage of these attributes include:

- Protective and corrective eyewear (e.g., eyeglass lenses and police/military visors)
- Sports safety equipment (e.g., helmets and protective eye/face visors)
- Automobiles (e.g., window glazing, sunroofs, impact-resistant bumpers)
- Compact discs and DVDs
- Life-saving medical devices (e.g., incubators, critical components of many medical devices that require clarity and can withstand sterilization)
- Food storage containers (e.g., reusable containers for food and beverages)
- Electronic equipment (e.g., casings for cell phones, computers, game consoles)
Polycarbonate has been used in many of these products for decades, it continues to be used today, and the global market for polycarbonate is growing for a very simple reason – it performs very well in a diverse range of products.

Epoxy resins are thermosets that have a unique combination of toughness, chemical resistance and strong adhesion. Most epoxy resins are made from BPA and these resins are widely used in products that we rely on every day. Epoxy resins are well suited to a wide range of coating applications, but also find growing use in high-strength/light weight composites. Common product categories where you’ll find epoxy resins include:

- Wind energy (e.g., wind turbine rotor blades)
- Electronics and electrical equipment (e.g., printed circuit boards, electrical equipment)
- Paints and coatings (e.g., water-based paints, protective coatings for metal cans and containers, powder coats for appliances, steel pipes, and primer coatings for automotive and marine applications, high-performance and decorative flooring)
- Aerospace (e.g., high-strength composites)
- Marine (e.g., manufacture and repair of boats)

As with polycarbonate, epoxy resins have been used in many of these products for 50 years or more, and new applications continue to be developed today due to the high performance attributes of epoxy resins. For both polycarbonate plastic and epoxy resins, it is not a simple matter to find a material that can match the attributes and performance of these materials.

Government Agencies Worldwide Support the Safety of BPA Based on a Strong Scientific Foundation

Arguably the most important factor to consider for alternatives to BPA is safety, since the controversy about BPA revolves entirely around its safety. Replacing BPA with an alternative that is not at least as safe would be nonsensical and potentially regrettable. It is not likely that any alternative has been so thoroughly tested or so comprehensively and frequently reviewed by government agencies as BPA. Regardless, the evidence supporting the safety of BPA speaks for itself and cannot be dismissed.

The most recent statement from a government agency on the safety of BPA is from the U.S. Food and Drug Administration (FDA) in June 2013. Its bottom line on safety is concisely and unambiguously captured in the form of a Q&A that appears on the FDA website:

"Is BPA safe? Yes. Based on FDA's ongoing safety review of scientific evidence, the available information continues to support the safety of BPA for the currently approved uses in food containers and packaging."

In further explanation, FDA states that its “current assessment is that BPA is safe at the very low levels that occur in some foods. This assessment is based on review by FDA scientists of hundreds of studies including the latest findings from new studies initiated by the agency.”

The views expressed by FDA are hardly unique. They are shared by many regulatory agencies that have each recently and independently evaluated the science on BPA. All have concluded that BPA is safe as currently used, including the reviews specific to materials in contact with food – which are likely to be the most sensitive applications. Other recent examples of government reviews include:

- European Food Safety Authority (January 2014)
- Hong Kong Centre for Food Safety (March 2013)
- Health Canada (September 2012)
FDA continues its robust review of BPA and has further stated:

“Because of concerns expressed in the last few years about the safety of BPA, FDA initiated additional studies to help determine whether or not BPA is safe as it is currently used in food packaging and containers. Some of these studies have been completed and others are on-going.”

“The FDA’s studies are being conducted by the agency’s National Center for Toxicological Research (NCTR). All are conducted under strict quality-assurance guidelines and are designed to produce information that will enhance our evaluation of BPA’s safety. The findings of the NCTR studies will be published in peer-reviewed scientific literature and will be reviewed by other experts including toxicologists and other scientists from the academic and private sectors as well as by other regulatory scientists. The results from these new studies so far support FDA’s assessment that the use of BPA in food packaging and containers is safe.”

Is There a Good Reason to Replace BPA?
In light of the strong scientific and government agency support for the safety of BPA, is there any good reason to replace BPA? If safety is the concern, FDA has already addressed it, and the simple answer is no. Replacement of BPA can only be defended for alternatives that, in fact, deliver better performance or are safer than BPA. Given the high performance of polycarbonate plastic and epoxy resins, and the long safety track record of BPA, those challenges will be very hard to meet.

For more information on BPA, polycarbonate plastic and epoxy resins, please visit http://www.FactsAboutBPA.org.

About Steve Hentges, Ph.D.
The author leads the Polycarbonate/BPA Global Group, which consists of the leading global manufacturers of BPA and polycarbonate plastic. This group of the American Chemistry Council sponsors health and environmental research and supports a wide range of communications and advocacy activities. Steve holds a Ph.D. in organic chemistry from Stanford University and a B.S. in chemistry from the University of Minnesota. In his current position, he has been deeply involved with the science on BPA for more than 14 years. The author periodically writes about the science of BPA in a Science 2.0 column at http://www.science20.com/steve_hentges.